

## 6.3 H-atom energy levels with spin-orbit coupling (Z=1)

$$\begin{aligned}
 E(n, \ell, s, j) &= -\frac{e^2}{2n^2 a_o (4\pi\epsilon_o)} + hc \alpha^2 R_\infty \frac{1}{2} \left\{ \frac{j(j+1) - \ell(\ell+1) - s(s+1)}{n^3 \ell \left(\ell + \frac{1}{2}\right) (\ell+1)} \right\} \\
 &= -\frac{e^2}{2n^2 a_o (4\pi\epsilon_o)} + hc \xi_{n,\ell} \frac{1}{2} [j(j+1) - \ell(\ell+1) - s(s+1)]
 \end{aligned}$$

**1s orbital:**  $n=1$   $\ell=0$   $s=\frac{1}{2}$   $j=\frac{1}{2}$  therefore, no SO coupling since  $l=0$

$$\therefore E_{1,0,\frac{1}{2},\frac{1}{2},m_j} = -\frac{e^2}{2a_o(4\pi\epsilon_o)} \quad :2 \text{ degenerate states with } m_j = \pm\frac{1}{2}$$

**2s orbital:**  $n=2$   $\ell=0$   $s=\frac{1}{2}$   $j=\frac{1}{2}$  therefore, no SO coupling since  $l=0$

$$\therefore E_{2,0,\frac{1}{2},\frac{1}{2},m_j} = -\frac{e^2}{8a_o(4\pi\epsilon_o)} \quad :2 \text{ degenerate states with } m_j = \pm\frac{1}{2}$$

**2p orbital:**  $n=2$   $\ell=1$   $s=\frac{1}{2}$  1.)  $j=\frac{3}{2}$ ;  $m_j = \pm\frac{3}{2}, \pm\frac{1}{2}$  :4 degenerate levels

$$\therefore E_{2,1,\frac{1}{2},\frac{3}{2},m_j} = -\frac{e^2}{8a_o(4\pi\epsilon_o)} + hc\xi_{2,1} \frac{1}{2} \left[ \underbrace{\left( \frac{3}{2} \right) \left( \frac{5}{2} \right) - (1)(2) - \left( \frac{1}{2} \right) \left( \frac{3}{2} \right)}_{=1} \right]$$

$$= -\frac{e^2}{8a_o(4\pi\epsilon_o)} + \frac{1}{2} hc\xi_{2,1}$$

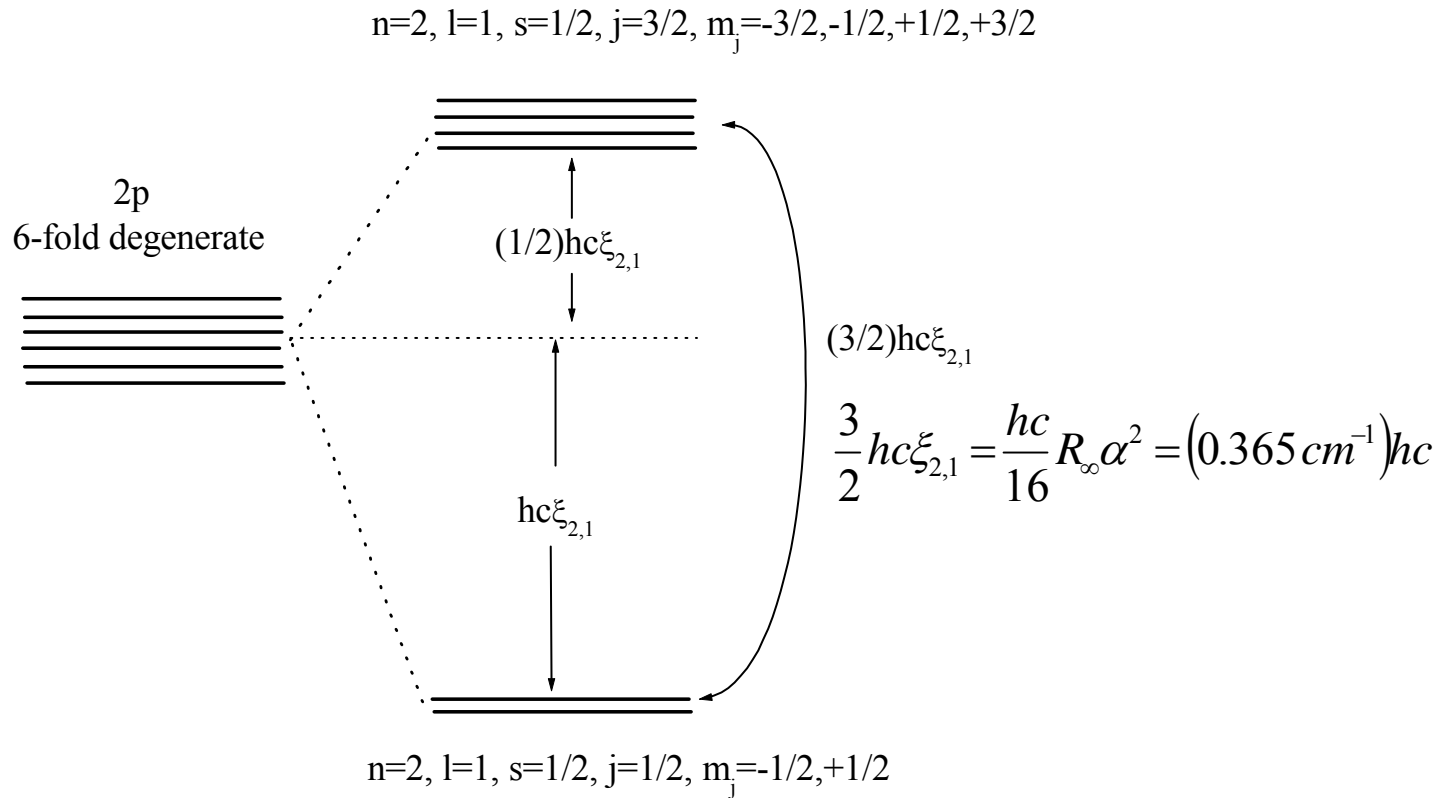
$$= -\frac{e^2}{8a_o(4\pi\epsilon_o)} + \frac{1}{2} hc\alpha^2 R_\infty \frac{1}{(8)(1)\left(\frac{3}{2}\right)(2)}$$

$$= -\frac{e^2}{8a_o(4\pi\epsilon_o)} + \frac{1}{3} \frac{hc}{16} \alpha^2 R_\infty$$

**2p orbital:**  $n=2$   $\ell=1$   $s=\frac{1}{2}$     2.)  $j=\frac{1}{2}$ ;  $m_j = \pm\frac{1}{2}$  :2 degenerate levels

$$\begin{aligned} \therefore E_{2,1,\frac{1}{2},\frac{1}{2},m_j} &= -\frac{e^2}{8a_o(4\pi\epsilon_o)} + hc\xi_{2,1} \frac{1}{2} \left[ \underbrace{\left( \frac{1}{2} \right) \left( \frac{3}{2} \right) - (1)(2) - \left( \frac{1}{2} \right) \left( \frac{3}{2} \right)}_{=-2} \right] \\ &= -\frac{e^2}{8a_o(4\pi\epsilon_o)} - hc\xi_{2,1} \\ &= -\frac{e^2}{8a_o(4\pi\epsilon_o)} - \frac{2}{3} \frac{hc}{16} \alpha^2 R_\infty \end{aligned}$$

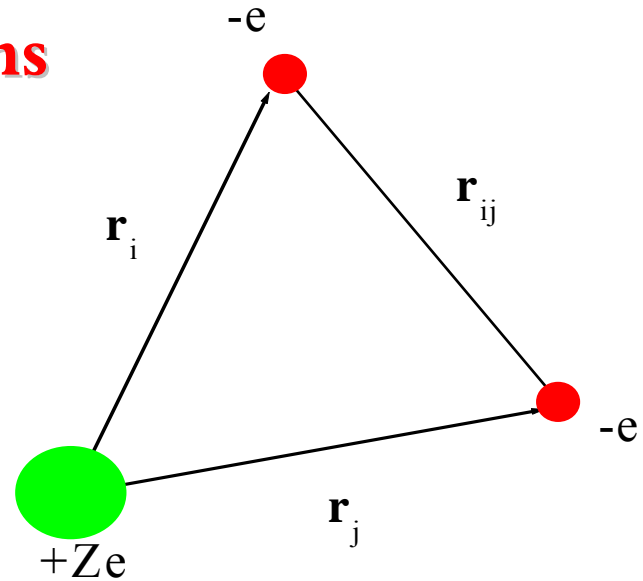
These 2 degenerate levels are lower in energy by  $hc\xi_{2,1}$



Compare with 
$$\Delta E_{n_2, n_1} = hcR_\infty \left\{ \frac{1}{n_1^2} - \frac{1}{n_2^2} \right\} = hc \cdot 0.75R_\infty = (82,303\text{ cm}^{-1})hc$$

Splitting is small (but measurable).

## 7: Many electron atoms



### 7.1: The Hamiltonian:

$$\hat{H} = \hat{T} + \hat{V} = \underbrace{-\frac{\hbar^2}{2m_e} \sum_{j=1}^N \vec{\nabla}_j^2}_{\text{kinetic energy}} + \underbrace{\sum_{j=1}^N \left( -\frac{Ze^2}{r_j} \right)}_{\text{Coulomb potential energy nucleus-electron attractive}} + \underbrace{\sum_{i>j}^N \frac{e^2}{r_{ij}}}_{\text{Coulomb potential energy electron-electron repulsive}}$$

## 7.2-i: The Angular Momenta: first scheme

$$\vec{L} = \sum_i \vec{\ell}_i \quad = \text{total orbital angular momentum}$$

$$\vec{S} = \sum_i \vec{s}_i \quad = \text{total spin angular momentum}$$

$$\vec{J} = \vec{L} + \vec{S} = \sum_i \vec{\ell}_i + \sum_i \vec{s}_i \quad = \text{total angular momentum}$$

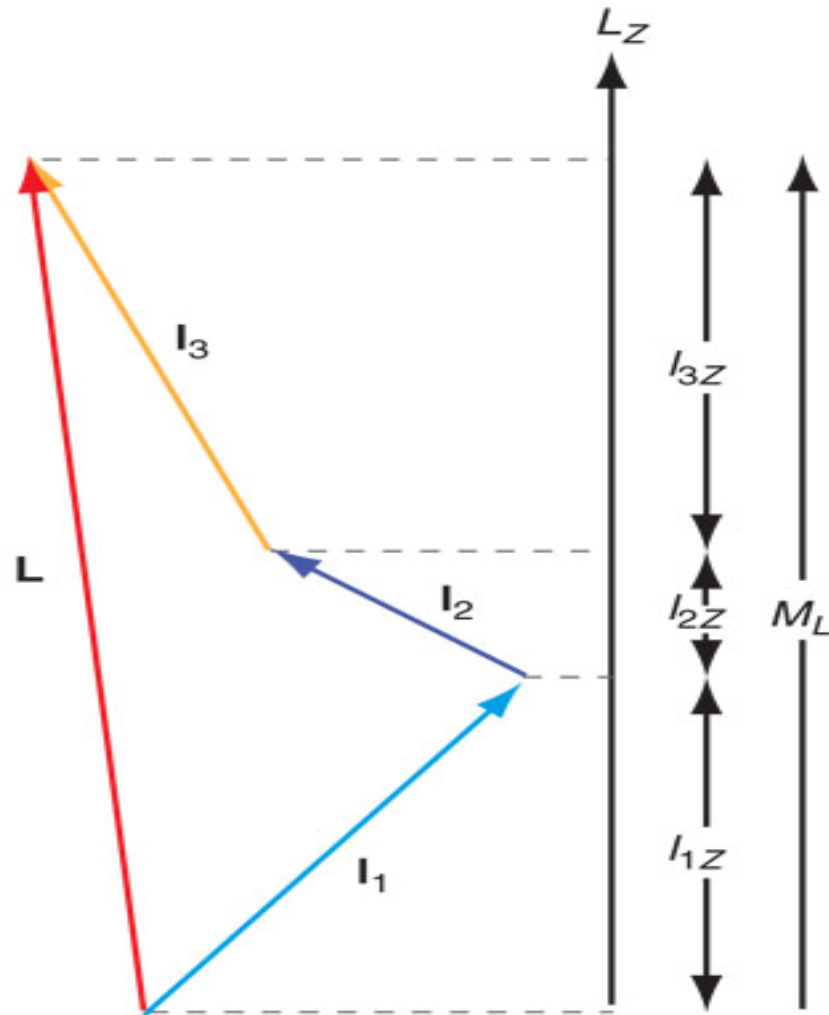
= Russell-Saunders or L-S coupling scheme  
= good for atom with  $Z \sim < 40$

$$\vec{L}_z = \sum_i \vec{L}_{iz} = \frac{\hbar}{i} \sum_i \left( x_i \frac{\partial}{\partial y_i} - y_i \frac{\partial}{\partial x_i} \right)$$

$$\vec{S}_z = \sum_i \vec{s}_{iz}$$

3 orbital angular momenta showing how

$$M_L = \sum_i m_{\ell_i}$$



## 7.2-ii: Angular Momentum: second scheme

$$\vec{j}_i = \vec{\ell}_i + \vec{S}_i \quad \text{for each electron}$$

$$\vec{J} = \sum_i \vec{j}_i$$

= j-j coupling scheme  
= important when spin-orbit interactions are large  
= important for heavier element  $Z \sim > 40$ .

## 7.3: Commutators: No LS coupling

$\hat{H}, \hat{L}^2, \hat{S}^2, \hat{L}_z, \hat{S}_z, \hat{\Pi}$  form a set of commuting, self-adjoint operators



| <u>Operator</u> | <u>Eigenvalues</u> |
|-----------------|--------------------|
| $\hat{H}$       | $E$                |
| $\hat{L}^2$     | $L(L+1)\hbar^2$    |
| $\hat{S}^2$     | $S(S+1)\hbar^2$    |
| $\hat{L}_z$     | $M_L\hbar$         |
| $\hat{S}_z$     | $M_S\hbar$         |
| $\hat{\Pi}$     | $I$                |

The wave functions depend on quantum numbers:  $E, L, S, M_L, M_S, I$   $\Rightarrow |E, L, S, M_L, M_S, I\rangle$

The wave functions are orthonormal so

$$\langle E, L, S, M_L, M_S, I | E', L', S', M_L', M_S', I' \rangle = \delta_{EE'} \delta_{LL'} \delta_{SS'} \delta_{M_L M_L'} \delta_{M_S M_S'} \delta_{II'}$$

and

$$\begin{aligned} \hat{L}_+ |E, L, S, M_L, M_S, I\rangle &= \sqrt{(L+M_L+1)(L-M_L)}\hbar |E, L, S, (M_L+1), M_S, I\rangle \\ \hat{L}_- |E, L, S, M_L, M_S, I\rangle &= \sqrt{(L-M_L+1)(L+M_L)}\hbar |E, L, S, (M_L-1), M_S, I\rangle \\ \hat{S}_+ |E, L, S, M_L, M_S, I\rangle &= \sqrt{(S+M_S+1)(S-M_S)}\hbar |E, L, S, M_L, (M_S+1), I\rangle \\ \hat{S}_- |E, L, S, M_L, M_S, I\rangle &= \sqrt{(S-M_S+1)(S+M_S)}\hbar |E, L, S, M_L, (M_S-1), I\rangle \end{aligned}$$

## 7.4: Eigenvalues, No LS coupling

a.)  $L = \text{integer} \geq 0$

$$\ell_1, \ell_2 \Rightarrow L' = \ell_1 + \ell_2, \ell_1 + \ell_2 - 1, \dots, |\ell_1 - \ell_2| \geq 0$$

$$L', \ell_3 \Rightarrow L = L' + \ell_3, L' + \ell_3 - 1, \dots, |L' - \ell_3|, \text{ etc.}$$

$$M_L = -L, -L + 1, \dots, L - 1, L$$

b.)  $S = \text{integer or half-integer} \geq 0$

$$N = 2: s_1, s_2 \Rightarrow S = 1, 0$$

$$N = 3: s_1, s_2, s_3 \Rightarrow S = \frac{3}{2}, \frac{1}{2} (\times 2)$$

$$N = 4: s_1, s_2, s_3, s_4 \Rightarrow S = 2, 1 (\times 3), 0 (\times 2)$$

Therefore if the number of electrons  $N$  is even  $S = \text{integer} \geq 0$ , and if the number of electrons  $N$  is odd  $S = \text{half-integer} \geq 0$

c.)  $E = E(L,S,I)$ :  $(2L+1)$  fold-degenerate in  $M_L$  and  $(2S+1)$  fold-degenerate in  $M_S$

### 7.5: Selection Rules: No LS coupling, electric-dipole induced,

$$\Delta L = \textcircled{0}, \pm 1$$

$$\Delta \ell = \pm 1$$

$$\Delta M_L = 0, \pm 1$$

$$\Delta m_\ell = 0, \pm 1$$

$$\Delta S = 0$$

Compare with those  
for H-like (one  
electron) systems:

$$\Delta s = 0$$

$$\Delta M_S = 0$$

$$\Delta m_s = 0$$

$$\Delta I \neq 0$$

$$\Delta I \neq 0$$

$\Delta \ell \neq 0$  for H-atom since  $I = (-1)^\ell$  and therefore parity forbids this.

In multielectron systems  $I \neq (-1)^L$  and so this restriction on L is gone.

Here parity is determined by an algebraic sum as: 
$$I = (-1)^{\sum_j \ell_j}$$

However, both states in transition cannot have  $L = 0, M_L = 0$

## 7.6: Commutators with LS coupling (Russell-Saunders)

$\hat{H}, \hat{L}^2, \hat{S}^2, \hat{J}^2, \hat{J}_z, \hat{\Pi}$  form a set of commuting, self-adjoint operators

| <u>Operator</u> | <u>Eigenvalue</u> |
|-----------------|-------------------|
| $\hat{H}$       | $E$               |
| $\hat{L}^2$     | $L(L+1)\hbar^2$   |
| $\hat{S}^2$     | $S(S+1)\hbar^2$   |
| $\hat{J}^2$     | $J(J+1)\hbar^2$   |
| $\hat{J}_z$     | $M_J\hbar$        |
| $\hat{\Pi}$     | $I$               |

The wave functions depend on quantum numbers:  $E, L, S, J, M_J, I$

$$\Rightarrow |E, L, S, J, M_J, I\rangle$$

The wave functions are orthonormal so

$$\langle E, L, S, J, M_J, I | E', L', S', J', M_{J'}, I' \rangle = \delta_{EE'} \delta_{LL'} \delta_{SS'} \delta_{JJ'} \delta_{M_J M_{J'}} \delta_{II'}$$

## 7.7: Eigenvalues, with LS coupling

$$L = \text{integer} \geq 0$$

$$S = \text{integer} \geq 0 \text{ for an even number of electrons}$$

$$S = \text{half-integer} \geq 0 \text{ for an odd number of electrons}$$

$$J = L + S, L + S - 1, \dots, |L - S| \quad \begin{array}{l} = \text{integer} \geq 0 \text{ if } S \text{ is an integer} \\ = \text{half-integer} \geq 0 \text{ if } S \text{ is a half-integer} \end{array}$$

$$M_J = -J, -J + 1, \dots, J - 1, J$$

## 7.8: Selection rules with LS coupling

$$\Delta J = 0, \pm 1$$

$$\Delta M_J = 0, \pm 1 \quad \text{But } J = 0 \rightarrow J' = 0 \text{ is forbidden.}$$

$$\Delta L = 0, \pm 1$$

$$\Delta S = 0$$

$$\Delta I \neq 0$$