## 8.2: An aside: Variational Theory, another approximation method

= an approximation method which works even when perturbation theory fails. Especially useful and important for many-electron systems.

**Theorem:** 
$$\widetilde{E} = \left\langle \widetilde{\psi} \mid \hat{H} \mid \widetilde{\psi} \right\rangle \geq E_o$$

Approximate energy of the system

"any" normalized function of the coordinates of the system including spin. = trial wave function which must satisfy the usual QM boundary conditions.

 $E_{o}$ 

 $\tilde{E}$ 

 $\widetilde{\psi}$ 

Lowest energy eigenvalue for the system (ground state)

**Proof:** 



(1) Set  $\widetilde{\psi} = \psi_{o}$ = exact ground state wave function.

$$\Rightarrow \widetilde{E} = \left\langle \psi_o \mid \widehat{H} \mid \psi_o \right\rangle = E_o \left\langle \psi_o \mid \psi_o \right\rangle = E_o$$

i.e. the theorem holds for  $\tilde{\psi} = \psi_o$  exactly

(2) When  $\tilde{\psi} \neq \psi_o$  which is usually the case

Will use our theorem that any arbitrary function can be expressed as an expansion in a complete basis set.

Use expansion postulate to expand  $\widetilde{\psi}$  in terms of a complete set of eigenfunctions of **H**.

$$\Rightarrow \widetilde{\psi} = \sum_{k=0}^{\infty} a_k \psi_k \quad \begin{bmatrix} \hat{H} \psi_k = E_k \psi_k; \langle \psi_\ell | \psi_k \rangle = \delta_{\ell k} \\ k = 0, 1, 2, \dots, \infty; E_o \leq E_1 \leq E_2 \dots \end{bmatrix}$$

Of course, if we actually new the exact wave functions of **H** then we would not be dealing with any approximation method. However, this is a formality we need to prove the theorem.

Require  $\widetilde{\psi}$  to be normalized

$$\therefore 1 = \left\langle \widetilde{\psi} \mid \widetilde{\psi} \right\rangle = \left\langle \sum_{k} a_{k} \psi_{k} \mid \sum_{p} a_{p} \psi_{p} \right\rangle$$
$$= \sum_{k} \sum_{p} a_{k}^{*} a_{p} \left\langle \psi_{k} \mid \psi_{p} \right\rangle = \sum_{k} \sum_{p} a_{k}^{*} a_{p} \frac{\delta_{kp}}{\sum_{p=1}^{n} if p \neq k}$$
$$= \sum_{k} a_{k}^{*} a_{k} = \sum_{k} \left| a_{k} \right|^{2}$$
Then
$$\widetilde{E} = \left\langle \widetilde{\psi} \mid \widehat{H} \mid \widetilde{\psi} \right\rangle = \left\langle \sum_{k} a_{k} \psi_{k} \mid \widehat{H} \mid \sum_{p} a_{p} \psi_{p} \right\rangle$$

$$=\sum_{k}\sum_{p}a_{k}^{*}a_{p}\left\langle\psi_{k}\mid\hat{H}\mid\psi_{p}\right\rangle=\sum_{k}\sum_{p}a_{k}^{*}a_{p}E_{p}\left\langle\psi_{k}\mid\psi_{p}\right\rangle$$

$$=\sum_{k}\sum_{p}a_{k}^{*}a_{p}E_{p}\delta_{kp}=\sum_{k}\left|a_{k}\right|^{2}E_{k}$$

Subtract E<sub>o</sub> from both sides

$$\widetilde{E} - E_o = \sum_{k=0}^{\infty} |a_k|^2 E_k - E_o = \sum_{k=0}^{\infty} |a_k|^2 E_k - E_o \sum_{k=0}^{\infty} |a_k|^2$$
$$= \sum_{k=0}^{\infty} |a_k|^2 (E_k - E_o)$$
But  $E_k \ge E_o$  for all k since  $E_o$  is the ground state energy. Note as well that

ground state energy. Note as well that  $|a_k|^2 \ge 0$  for all k.

This theorem is the basis of the variational method for calculating approximate wave functions and energies of many electron atoms and molecules

**Note:** The theorem holds for the lowest energy state for each state of a given symmetry, if

 $\widetilde{\psi}$  is chosen to have the correct symmetry; for example, the lowest s state, p state, d state etc.

$$\Rightarrow \widetilde{E} \geq E_{sym,\ell}$$

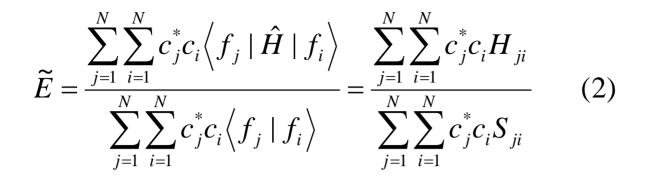
## **Variational Method**

Start with 
$$\hat{H}\widetilde{\psi} = \widetilde{E}\widetilde{\psi}$$

Premultiply by  $\widetilde{\psi}^*$  integrate, and rearrange for E

$$\Rightarrow E = \frac{\left\langle \tilde{\psi} \mid \hat{H} \mid \tilde{\psi} \right\rangle}{\left\langle \tilde{\psi} \mid \tilde{\psi} \right\rangle}$$
(1)  
Let:  $\tilde{\psi} = \sum_{i=1}^{N} c_i f_i$ 

where  $\{f_i\}$  are functions that satisfy the general conditions for a wave function. They may be constitute an orthonormal set or they may not.



where

$$H_{ji} = \left\langle f_{j} \mid \hat{H} \mid f_{i} \right\rangle \; ; \; S_{ji} = \left\langle f_{j} \mid f_{i} \right\rangle$$

 $H_{ji}$  is a Hamiltonian matrix element and  $S_{ji}$  is an overlap integral

Rearrange (2) such that

$$\widetilde{E}\sum_{j=1}^{N}\sum_{i=1}^{N}c_{j}^{*}c_{i}S_{ji} = \sum_{j=1}^{N}\sum_{i=1}^{N}c_{j}^{*}c_{i}H_{ji} \qquad (3)$$

We want coefficients  $\{c\}$  that make E a minimum. We can do this by differentiating (3) with respect to  $c_k$  and setting  $\frac{\partial E}{\partial E}$ 

$$\frac{\partial E}{\partial c_k} = 0$$

Do the case for i,j = 1,2

$$\widetilde{E}(c_1^2S_{11}+2c_1c_2S_{12}+c_2^2S_{22})=c_1^2H_{11}+2c_1c_2H_{12}+c_2^2H_{22}$$

Differentiate with respect to  $c_1$ 

$$(2c_1S_{11} + 2c_2S_{12})E + (c_1^2S_{11} + 2c_1c_2S_{12} + c_2^2S_{22})\frac{\partial E}{\partial c_1}$$
  
=  $2c_1H_{11} + 2c_2H_{12}$ 

Similarly, differentiation with respect to  $c_2$  yields:

$$(2c_2S_{22} + 2c_1S_{12})E + (c_1^2S_{11} + 2c_1c_2S_{12} + c_2^2S_{22})\frac{\partial E}{\partial c_2}$$
  
=  $2c_2H_{22} + 2c_1H_{12}$ 

Set 
$$\frac{\partial E}{\partial c_1} = \frac{\partial E}{\partial c_2} = 0$$
 and rearrange to show:  
 $(H_{11} - ES_{11})c_1 + (H_{12} - ES_{12})c_2 = 0$   
 $(H_{12} - ES_{12})c_1 + (H_{22} - ES_{22})c_2 = 0$ 

This will have non-trivial  $\{c_i\}$  solutions if the following determinant = 0.

$$\begin{vmatrix} H_{11} - ES_{11} & H_{12} - ES_{12} \\ H_{21} - ES_{21} & H_{22} - ES_{22} \end{vmatrix} = 0$$

= **SECULAR EQUATION** (like before)

For N basis functions, the secular equation is given by:

$$\begin{vmatrix} H_{11} - ES_{11} & H_{12} - ES_{12} & \dots & H_{1N} - ES_{1N} \\ H_{21} - ES_{21} & H_{22} - ES_{22} & \dots & H_{2N} - ES_{2N} \\ \vdots & \vdots & \vdots & \vdots \\ H_{N1} - ES_{N1} & H_{N2} - ES_{N2} & \dots & H_{NN} - ES_{NN} \end{vmatrix} = 0$$

The secular equation yields a N<sup>th</sup> order polynomial in E. For each E can solve for  $\{c_i\}$  to get the "best" approximate wave function for the system having that E.

Back to the problem:

## 8.3: Optimization of the Radial Part by the Variation Method

What function should we use for the individual **entries**,  $\phi_i(k)$  in the determinant? Instead of the H-atom eigenfunctions, we use modified functions. The key difference in the modified functions from the H-atom functions

## Effective nuclear charge Zeta ( $\zeta$ ) < the true nuclear charge

The outermost electrons are **shielded** from the full nuclear charge by other electrons. Each 1-e<sup>-</sup> orbital is constructed from a linear combination of H-atom-like orbitals. All coefficients in the linear combination in each orbital are used as variational parameters. The  $\zeta$  values are optimized separately.

**Next:** solve the Schrodinger Equation for each electron *i*. To do so, we must know  $V_i^{eff}$ . This means that we must know the functional forms of all the other orbitals, also the case for the remaining n-1 electrons. That is, the answers must be known to solve the problem.

The way out of this quandary is to use an iterative approach.

- Make a reasonable guess for an initial set of orbitals
- Calculate an effective potential using these orbitals
- Calculate the energy and orbital functions for each of the n electrons in turn
- New orbitals are used to refine the initial guesses

Repeat until the solutions for the energies and the orbitals are self-consistent