

1

H

Hydrogen

1.007 94

2

He

Helium

4.002 602

3

Li

Lithium

6.941

4

Be

Beryllium

9.012 182

5

B

Boron

10.811

6

C

Carbon

12.0107

7

N

Nitrogen

14.006 74

8

O

Oxygen

15.999 4

9

F

Fluorine

18.998 4032

10

Ne

Neon

20.1797

11

Na

Sodium

22.989 770

12

Mg

Magnesium

24.3050

13

Al

Aluminum

26.981 538

14

Si

Silicon

28.0855

15

P

Phosphorus

30.973 761

16

S

Sulfur

32.066

17

Cl

Chlorine

35.4527

18

Ar

Argon

39.948

19

K

Potassium

39.0983

20

Ca

Calcium

40.078

21

Sc

Scandium

44.955 910

22

Ti

Titanium

47.867

23

V

Vanadium

50.9415

24

Cr

Chromium

51.9961

25

Mn

Manganese

54.938 049

26

Fe

Iron

55.845

27

Co

Cobalt

58.933 200

28

Ni

Nickel

58.6934

29

Cu

Copper

63.546

30

Zn

Zinc

65.39

31

Ga

Gallium

69.723

32

Ge

Germanium

72.61

33

As

Arsenic

74.921 60

34

Se

Selenium

78.96

35

Br

Bromine

79.904

36

Kr

Krypton

83.80

37

Rb

Rubidium

85.4678

38

Sr

Strontium

87.62

39

Y

Yttrium

88.905 85

40

Zr

Zirconium

91.224

41

Nb

Niobium

92.906 38

42

Mo

Molybdenum

95.94

43

Tc

Technetium

(98)

44

Ru

Ruthenium

101.07

45

Rh

Rhodium

102.905 50

46

Pd

Palladium

106.42

47

Ag

Silver

107.8682

48

Cd

Cadmium

112.411

49

In

Indium

114.818

50

Sn

Tin

118.710

51

Sb

Antimony

121.760

52

Te

Tellurium

127.60

53

I

Iodine

126.904 47

54

Xe

Xenon

131.29

55

Cs

Cesium

132.905 45

56

Ba

Barium

137.327

57

La

Lanthanum

138.9055

72

Hf

Hafnium

178.49

73

Ta

Tantalum

180.9479

74

W

Tungsten

183.84

75

Re

Rhenium

186.207

76

Os

Osmium

190.23

77

Ir

Iridium

192.217

78

Pt

Platinum

195.078

79

Au

Gold

196.966 55

80

Hg

Mercury

200.59

81

Tl

Thallium

204.3833

82

Pb

Lead

207.2

83

Bi

Bismuth

208.980 38

84

Po

Polonium

(209)

85

At

Astatine

(210)

86

Rn

Radon

(222)

87

Fr

Francium

(223)

88

Ra

Radium

(226)

89

Ac

Actinium

(227)

104

Rf

Rutherfordium

(261)

105

Db

Dubnium

(262)

106

Sg

Seaborgium

(263)

107

Bh

Bohrium

(264)

108

Hs

Hassium

(265)[†]

109

Mt

Meitnerium

(268)[†]

110

Uun*

Ununnilium

(269)[†]

111

Uuu*

Unununium

(272)[†]

112

Uub*

Unbibium

(277)[†]

114

Uuq*

Ununquadium

(285)[†]

58

Ce

Cerium

140.116

59

Pr

Praseodymium

140.907 65

60

Nd

Neodymium

144.24

61

Pm

Promethium

(145)

62

Sm

Samarium

150.36

63

Eu

Europium

151.964

64

Gd

Gadolinium

157.25

65

Tb

Terbium

158.925 34

66

Dy

Dysprosium

162.50

67

Ho

Holmium

164.930 32

68

Er

Erbium

167.26

69

Tm

Thulium

168.934 21

70

Yb

Ytterbium

173.04

71

Lu

Lutetium

174.967

90

Th

Thorium

232.0381

91

Pa

Protactinium

231.035 88

92

U

Uranium

238.0289

93

Np

Neptunium

(237)

94

Pu

Plutonium

(244)

95

Am

Americium

(243)

96

Cm

Curium

(247)

97

Bk

Berkelium

(247)

98

Cf

Californium

(251)

99

Es

Einsteinium

(252)

100

Fm

Fermium

(257)

101

Md

Mendelevium

(258)

102

No

Nobelium

(259)

103

Lr

Lawrencium

(262)

Period

1

2

3

4

5

6

7

Key:

6

Atomic number

C

Symbol

Carbon

Name

12.0107

Average atomic mass

Metals

Alkali metals

Alkaline-earth metals

Transition metals

Other metals

Nonmetals

Hydrogen

Semiconductors

Halogens

Noble gases

Other nonmetals

A team at Lawrence Berkeley National reported the discovery of elements 116 and 118 in June 1999. The same team retracted the discovery in July 2001. The discovery of element 114 has been reported but not confirmed.

Key:

6 — Atomic number
C — Symbol
Carbon — Name
12.0107 — Average atomic mass

Metals

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- Alkaline-earth metals
- Transition metals
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Nonmetals

- Hydrogen
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A team at Lawrence Berkeley National reported the discovery of elements 116 and 118 in June 1999. The same team retracted the discovery in July 2001. The discovery of element 114 has been reported but not confirmed.

[†] Estimated from currently available IUPAC data.

* The systematic names and symbols for elements greater than 109 will be used until the approval of trivial names by IUPAC.

The atomic masses listed in this table reflect the precision of current measurements. (Values listed in parentheses are those of the element's most stable or most common isotope.) In calculations throughout the text, however, atomic masses have been rounded to two places to the right of the decimal.